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Key indicators

Single-crystal X-ray study

T = 295 K

Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$

R factor = 0.069

wR factor = 0.218

Data-to-parameter ratio = 14.6

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2,2'-(Ethane-1,2-diyl)dibenzimidazolium
bis(perchlorate) tetrahydrate

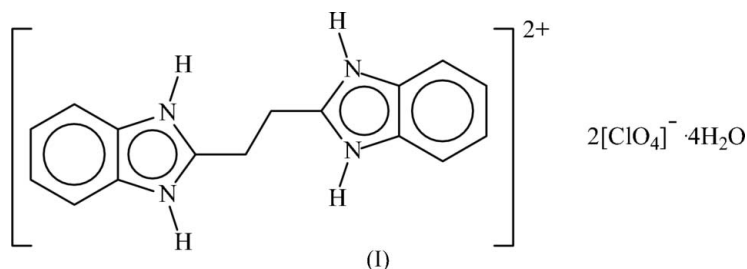
The cation, anion and water molecules of the title compound, $\text{C}_{16}\text{H}_{16}\text{N}_4^{2+} \cdot 2\text{ClO}_4^- \cdot 4\text{H}_2\text{O}$, are linked by hydrogen bonds into a three-dimensional network. The cation lies on a center of inversion, which is located at the mid-point of the $-\text{CH}_2\text{CH}_2-$ bond.

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Comment

The 2,2'-(ethane-1,2-diyl)dibenzimidazole nitrogen heterocycle, whose synthesis has been described (Dudd *et al.*, 2003), affords crystalline compounds with metal salts; with some, the nitrogen heterocycle binds to the metal atom (Albada *et al.*, 1999, 2000), whereas in others it does not bind, existing instead as a protonated dication (Broughton *et al.*, 1998; Matthews *et al.*, 2003). An attempt to synthesize an erbium perchlorate complex gave 2,2'-(ethane-1,2-diyl)dibenzimidazolium bis(perchlorate) tetrahydrate, (I) (Fig. 1). The cation lies on a center of inversion, and its four nitrogen-bound H atoms are engaged in hydrogen bonding with the water molecules. The cation, anion and water molecules are linked by hydrogen bonds (Table 1) into a three-dimensional network.



Experimental

To a hot solution of 2,2'-(ethane-1,2-diyl)dibenzimidazole (0.07 g, 0.30 mmol) in ethanol (5 ml) was added a solution of erbium(III) perchlorate (0.34 g, 0.60 mmol) in acetonitrile (5 ml), giving a clear light-red solution. The solution was filtered and pale-yellow crystals were isolated from the filtrate after 10 d.

Crystal data

 $\text{C}_{16}\text{H}_{16}\text{N}_4^{2+} \cdot 2\text{ClO}_4^- \cdot 4\text{H}_2\text{O}$ $M_r = 535.29$ Monoclinic, $P2_1/n$ $a = 7.0898 (6) \text{ \AA}$ $b = 17.204 (2) \text{ \AA}$ $c = 9.9894 (9) \text{ \AA}$ $\beta = 110.072 (2)^\circ$ $V = 1144.4 (2) \text{ \AA}^3$ $Z = 2$ $D_x = 1.553 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.35 \text{ mm}^{-1}$ $T = 295 (2) \text{ K}$

Block, pale yellow

 $0.20 \times 0.16 \times 0.11 \text{ mm}$

Data collection

Bruker APEX-II area-detector diffractometer
 φ and ω scans
 Absorption correction: none
 7440 measured reflections

2601 independent reflections
 1988 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$
 $\theta_{\text{max}} = 27.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.069$
 $wR(F^2) = 0.218$
 $S = 1.05$
 2601 reflections
 178 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1223P)^2 + 0.991P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.99 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.48 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O1w-H1w1 \cdots O1$	0.84 (1)	2.13 (2)	2.915 (4)	154 (4)
$O1w-H1w2 \cdots O2w^i$	0.84 (1)	2.29 (2)	3.083 (4)	158 (4)
$O2w-H2w1 \cdots O1^{ii}$	0.85 (1)	2.11 (2)	2.907 (4)	157 (5)
$O2w-H2w2 \cdots O1w^{iii}$	0.85 (1)	2.13 (4)	2.897 (4)	150 (7)
$N1-H1n \cdots O1w$	0.85 (1)	2.02 (1)	2.853 (4)	169 (4)
$N2-H2n \cdots O2w$	0.85 (1)	1.96 (1)	2.804 (4)	172 (3)

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$; (iii) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$.

The displacement parameters of the O atoms of the perchlorate were restrained to be nearly isotropic. The carbon-bound H atoms were placed at calculated positions ($C-H = 0.93$ and 0.97 \AA) and were included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H})$ set at $1.2U_{\text{eq}}(\text{C})$. The water and amino H atoms were located in a difference Fourier map, and were refined with distance restraints of $O-H = N-H = 0.85$ (1) \AA and $H \cdots H = 1.39$ (1) \AA ; their displacement parameters were freely refined.

Data collection: *APEX-2* (Bruker, 2006); cell refinement: *SAINT* (Bruker, 2006); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-*

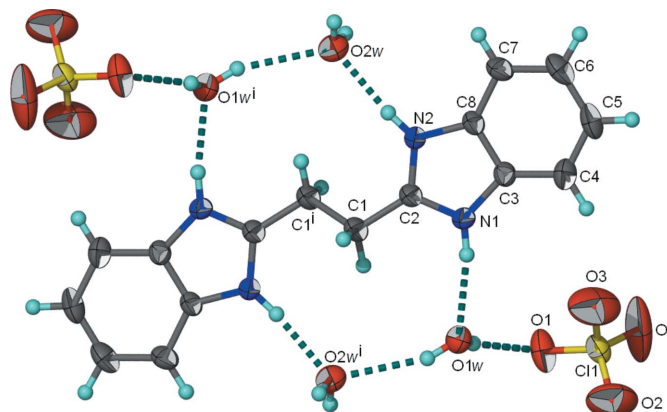


Figure 1

The formula unit of (I), with displacement ellipsoids drawn at the 50% probability level. H atoms are drawn as spheres of arbitrary radius. [Symmetry code: (i) $1 - x, 1 - y, 1 - z$.]

SEED (Barbour, 2001); software used to prepare material for publication: *SHELXL97*.

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